

State-of-the-art, opportunities and challenges for nanotechnology in concrete materials

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Cement and concrete

Cement and nanotechnology

Cement-based material and nanotechnology

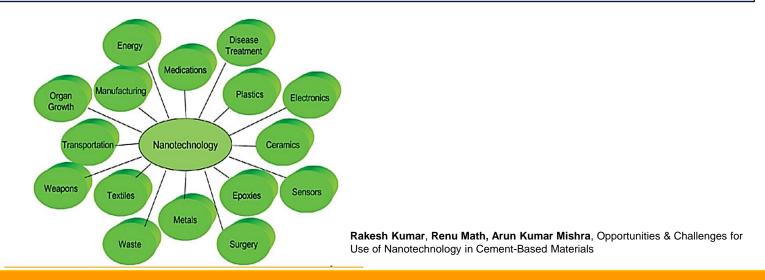
Cement-based materials and carbon nanotechnology

Cement and nanotechnology

Nanotechnology

- ~30 years old: -1959, *R.Feyman*n, building things from the bottom up with atomistic precision
 - 1974, N. Taniguchi, first use of the term
 - -1980s, *K.E Drexler*, popularization and awareness of nanotechnology concepts *G.Binning and H.Roher*, scanning tunnelling microscope invention
- Nanoparticles: from 1 to100nm (National Nanotechnology Initiatives, U.S)

Nanotechnology is the understanding and control of matter at dimensions between approximately 1 and 100 nanometers, where unique phenomena enable novel applications (EU commission).



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Cement and nanotechnology

Nanotechnology applied to cementitious materials:

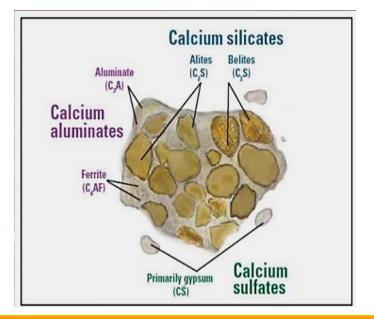
1817 L.Vicat: artificial cement by burning an intimate mixture of chalk and clay **1824 J.Aspdin**: proto-Portland cement

To make cement and concrete a product of nanotechnology we must be able to **understand and control** type, amount, structure and location of the:

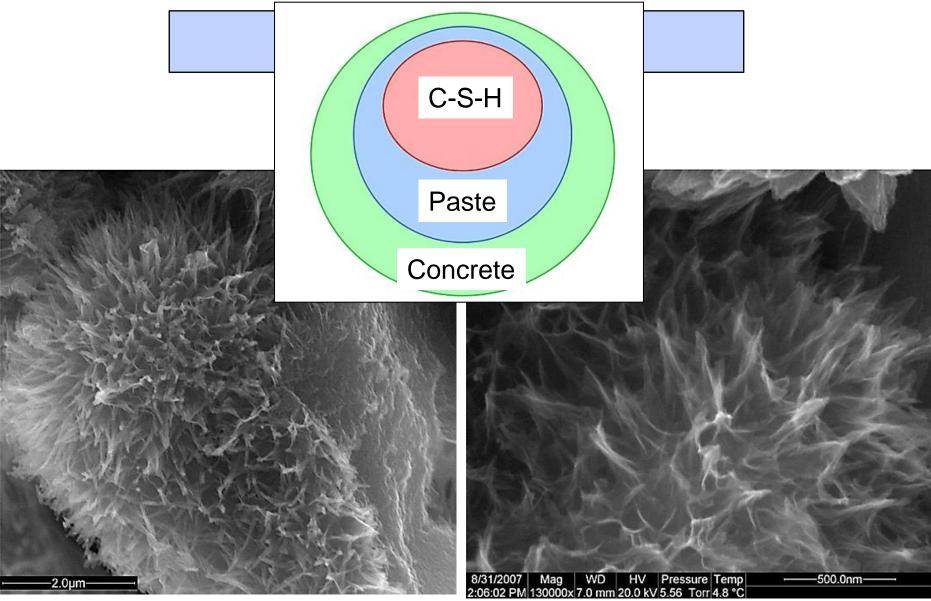
- nano-products
- nano-scale pores
- nano-ingredients

to create materials with fundamentally new properties and functions.

Anhydrous cement Calcium aluminate -tricalcium aluminate C₃A, 8-12% bwc -calcium aluminoferite C₄AF, 6-12% bwc Calcium silicate -tricalcium silicate C₃S, 55-65% bwc -dicalcium silicate C₂S, 15-25% bwc Calcium sulfates -gypsum or hemihydrate



Hydrated cement: C-S-H

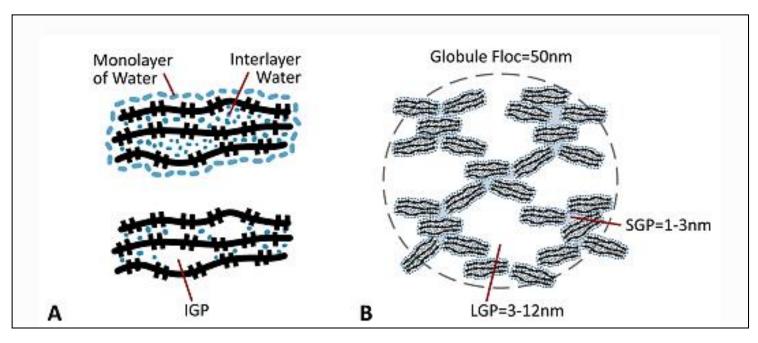


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Hydrated cement: C-S-H

J-model for C-S-H (2008)

Variable, nanoscale composite material itself, based on colloidal particle with a layered substructure modifyed by a multinetwork of hierarchycal capillary pores and cracks



- a. Particles >5nm*30-60nm outer surface and internal porosity
- b. Gel pores (IGP intragranular, SGP small, LGP large gel pores)
- c. Clusters of globules: HD C-S-H, LD C-S-H

Resolving C-S-H

C-S-H

- nanoscale dimensions;
- the most **abundant** reaction product, ~50% of paste volume;
- not intrinsically strong or stable;
- 70% LD C-S-H: 21.7GPa; 30% HD C-S-H: 29.4GPa (Constantinides and Ulm, CCR 34,2004)
- forming a continuous layer binding together the original cement particles into a cohesive whole
- all the other hydration products being discrete crystals not forming strong connections to the solid phases they are in contact with and so not contributing much to the global strength.

"Understanding" C-S-H should open the door to the potential "control" of fundamental and engineering properties:

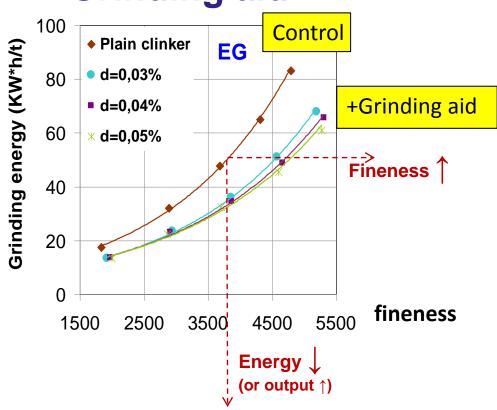
- tensile strength;
- shrinkage cracking potential;
- ions substitution/interactions (e.g AI) and effect on bulk properties;
- chemical compatibility with inorganic material additions (e.g fly-ash);
- permeability;
- in depth structural modifications

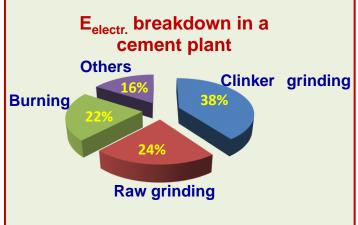
Challenge !!!

Anhydrous clinker: grinding (1)

- Clinker grinding: ~40% (of electrical plant demand)
- Organic compounds (GA): able to reduce energy consumption (glycols and amines)

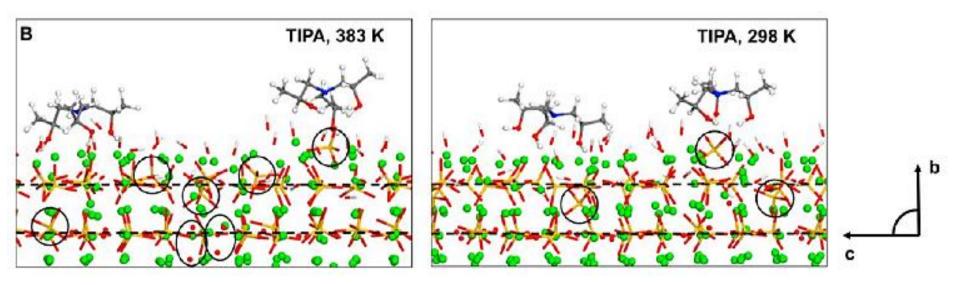
Grinding aid





Anhydrous clinker: grinding (2)

- Design and prediction behaviour: need adsorption and agglomeration energy values
- Recent progress by computer-aided simulation



Mishra R.K, Flatt R.J, Heinz H., J.Phys.Chem. C 2013,117, 104127/10432

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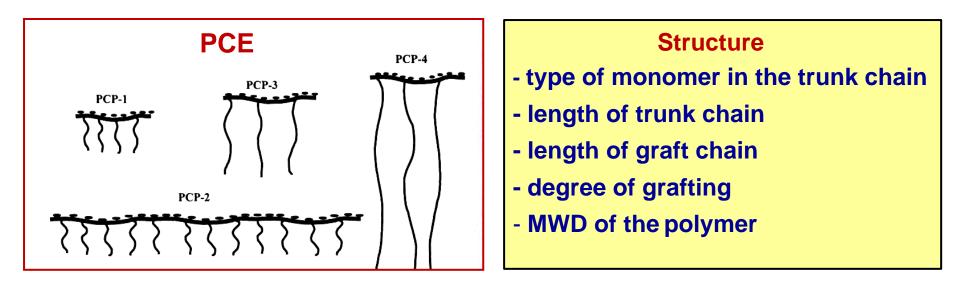
Hydrated cement: paste dispersion (1)

Superplasticizer

- Paste initial dispersibility
- Paste dispersibility over time
- Mixture with low water added



Hydrated cement: dispersion (2)



Fundamental components of comb-type SPs :

-carboxylic groups of the trunk chain, as adsorbing sites, bonded to Ca⁺⁺ on the cement particles

-polyether graft chains able to reduce attractive forces between the cement particles by their steric hindrance effect.

Cement and concrete

- Cement and nanotechnology
- Cement-based material and nanotechnology

Cement-based materials and carbon nanotechnology

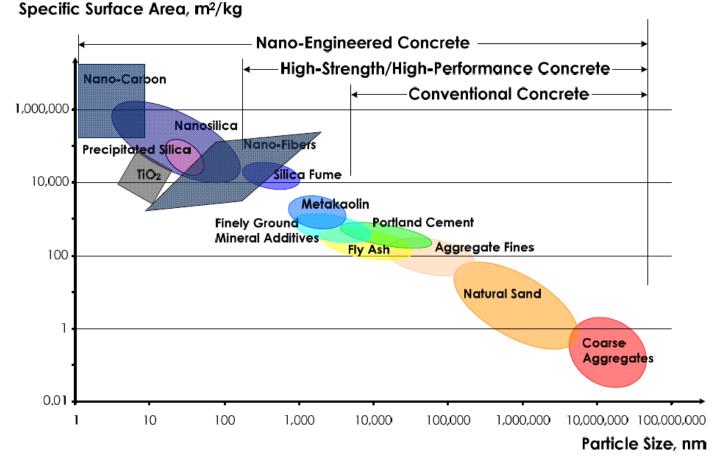
Additions of nanosized materials (1)

Main nanoparticles

- colloidal SiO₂
- TiO₂
- nano-CaCO₃

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- **nano-Al₂O₃** (fibers)



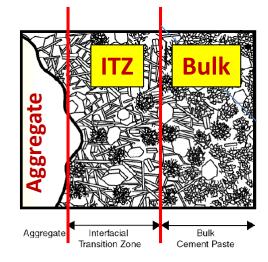
Additions of nanosized materials (2)

Nanoscale particles: high surface area-to-volume ratio high reactivity (some)

Formation of denser microstructure

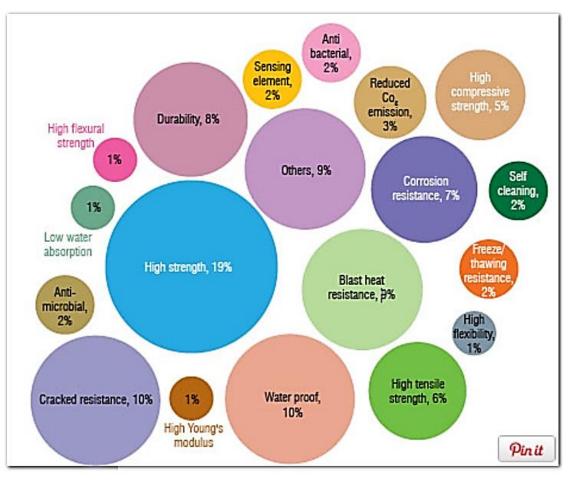
- nanoparticles act as centers of crystallization of hydration products
- filling the nanosize pores of cement paste
- improve structure of the aggregate contact zone (ITZ)
- generation more C-S-H through pozzolanic reaction

Pozzolan + CH + $H \xrightarrow{\text{SLOW}} C-S-H$



- favour formation of small-sized crystal ((CaOH)2 and Afm) and small-sized clusters C-S-H
- arrest crack and create interlock effect between slip planes

Additions of nanosized materials (3)



Percentage-wise distribution of patents based on characteristic property imparted by the added nanomaterials. *NanoWerk spotlight, dec. 2012*

Advantages

- mechanical properties \uparrow
- durability \uparrow
- reactivity ↑
- industrial footprint ↓

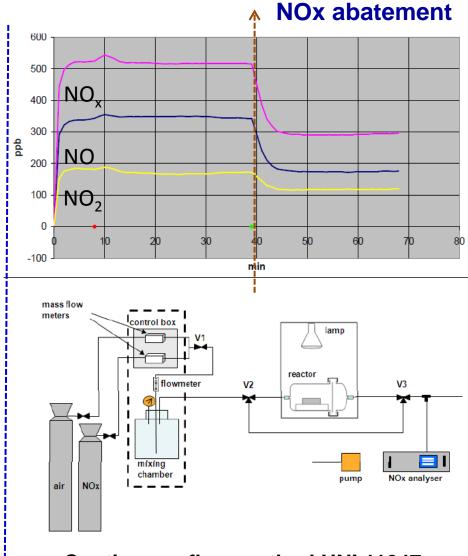
TiO₂: de-polluting property

Photocatalytic cement

TiO2: effective in pollutants reduction (NOx, aromatics, ammonia and aldehydes)







Continuous flow-method UNI 11247

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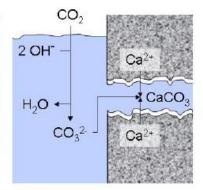
TiO₂: self-cleaning property



Self-healing materials

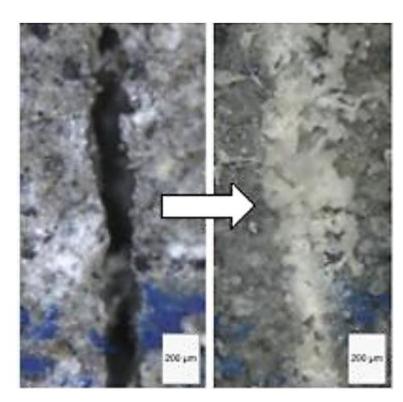
a. Autogeneous healing

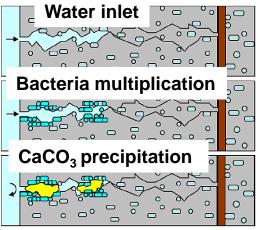
 $CO_2 + Ca(OH)_2 \rightarrow CaCO_3 + H_2O$



- b. Healing by compound expansion
- c. Healing by compound swelling
- d. Bacteria

bacteria Lactate + $O_2 \rightarrow$ acetate + $CaCO_3 + CO_2$ Metabolic conversion of Calcium Lactate ----> $CO_2^{Ca(OH)_2}$ ---->CaCO3





After Klaas Van Breugel

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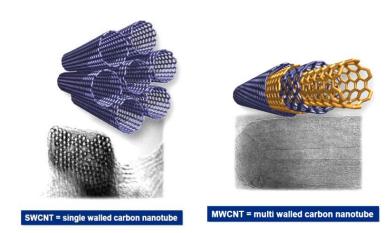
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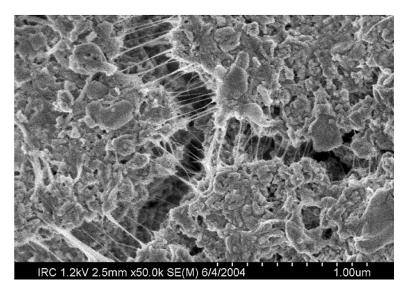
Cement-based materials and carbon nanotechnology

CNTs (1)

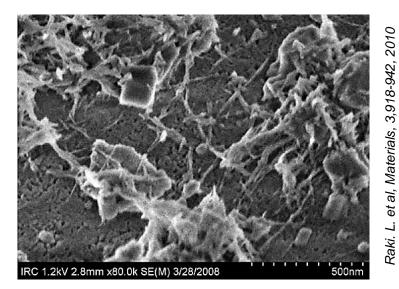
SWCNTs: 0.4-10nm diameterMWCNTs: 10-100nm diameterAspect ratio: 30 - many thousands



Nano reinforcement of cement-based composites (E~1TPa, Tensile strength~10-50GPa) **Potential:** crack propagation resistance (finely diistribution throughout cement matrix, bridges across voids...) \uparrow , strength \uparrow , stiffness \uparrow , ductility \uparrow



Crack bridging in a SWCNT/hydrated OPC composite



Growth of C-S-H around SWCNT bundles at 135 min. of hydration of an OPC composite sample

CNTs (2)

Dispersibility in cementitious matrix: hydrofobicity

- surfactants (comb-polymer)
- functionalization (-COOH, -OH)
- synthesis on cement grain surface
- sonication

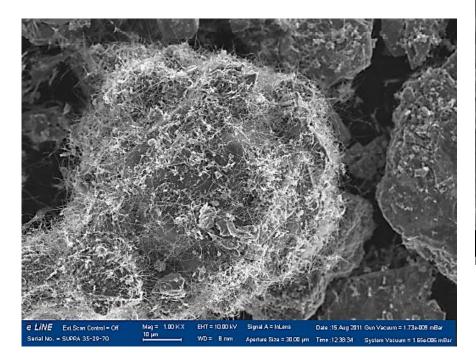
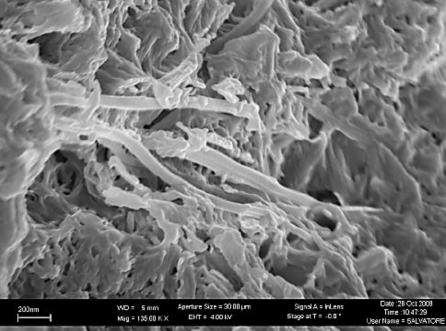


Figure 1: CNT synthesized on the surface of a cement grain. The image is $115 \,\mu\text{m}$ in width. Image by K. Hruška, Institute of Physics ASCR, Prague.



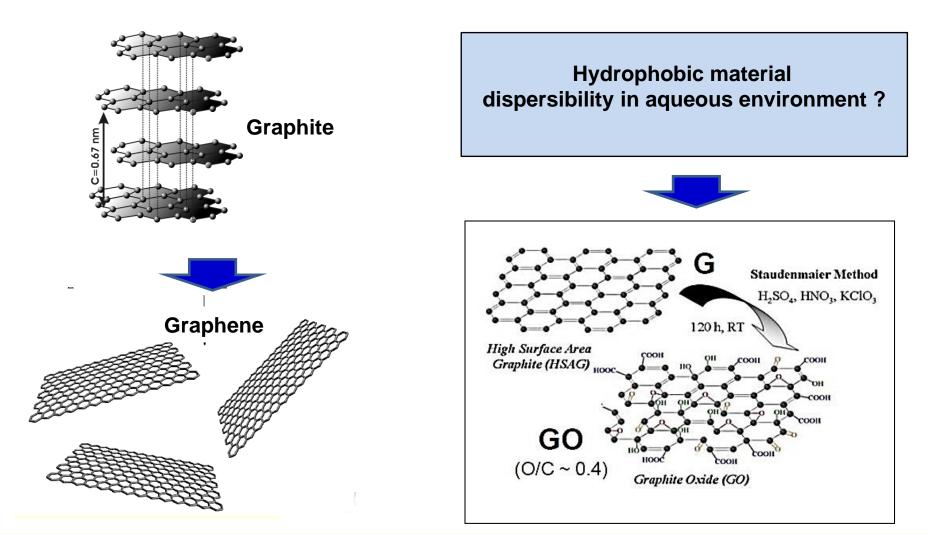
SEM micrograph of 0.4% MWCNT cement composite

Musso S. et al., Composites Science and Technology, 69 (2009), 1985

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Graphene (1)

One-atom thick planar sheet of sp2-bonded carbon atoms, densely packed in a honeycomb crystal lattice. Graphite itself consisting of many graphene sheets stacked together.



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Graphene (2)

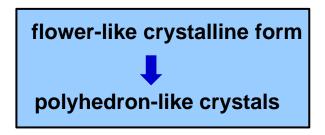
GO interaction with cement

a. Microstructure

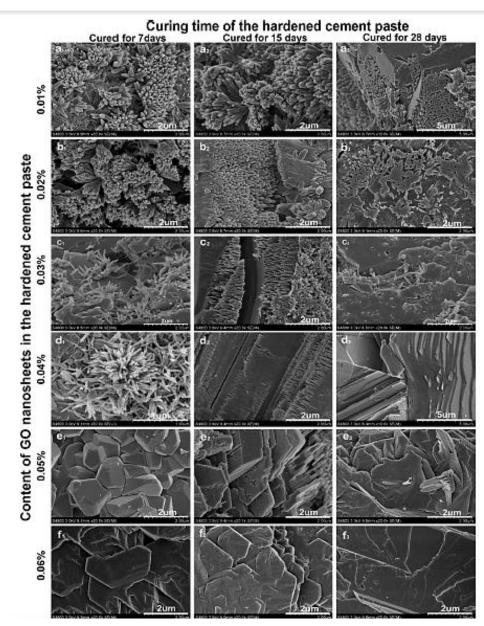
7d: flower-like crystalline form

15d: clear tendency for aggregates of flower-like and rod-like crystals to form a planar structure

28d: a compact structure is formed, and rod-like crystals can hardly be seen in hardened cement paste with any GO content



Shengua Lv et al., CrystEngComm., 2014, 16, 8508



SEM images of fractured surface of hardened cement paste

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Graphene (3)

GO interaction with cement b. Porosity

GO content (wt/wt%)	Porosity (%)
0.01	31.84
0.02	23.56
0.03	15.41
0.04	12.62
0.05	9.38
0.06	6.24

1

c. Strength

Table 5 Flexural and compressive strength and elastic modulus of hardened cement pastes with different GO contents

GO content (wt/wt%)	Compressive strength (MPa)/rate of increase (%)		Flexural strength (MPa)/rate of increase (%)	
	7 days	28 days	7 days	28 days
0 (control sample)	36.74/0	59.31/0	5.63/0	8.84/0
0.01	45.31/23.3	69.65/17.4	8.85/57.2	12.34/39.6
0.02	49.51/34.75	77.82/31.2	9.21/63.6	13.68/54.8
0.03	55.56/51.22	86.62/46.1	9.93/76.4	14.72/66.5
0.04	58.61/59.5	92.36/55.7	9.88/75.5	14.74/67.1
0.05	62.25/69.4	93.38/57.4	9.37/66.4	13.53/53.1
0.06	63.27/72.7	94.26/58.5	9.19/63.2	12.63/42.9

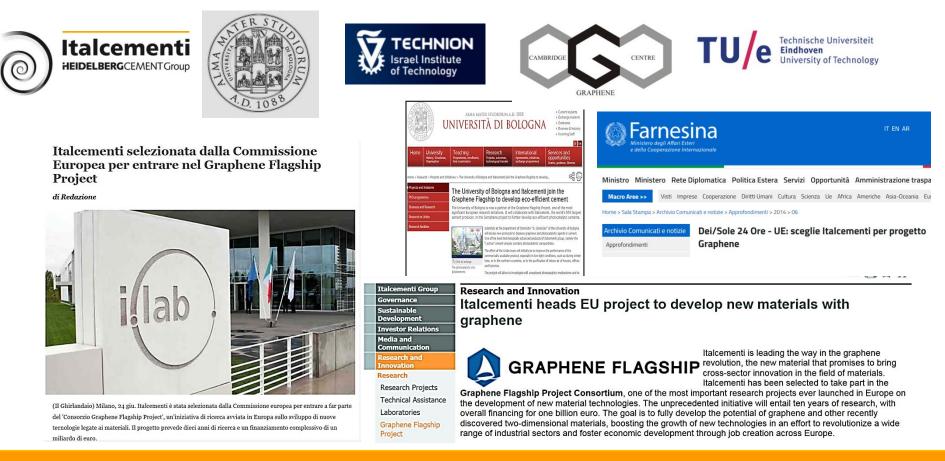
Carbon nanotechnology: ITC and Graphene

10/2014-03/2016: ramp-up phase

Task: T10.12 of WP 10, Composites for catalysis in building applications

04/2016-03/2018: core 1 project

Task: T 13.3.4.12 of WP 13, Coatings for photocatalytic applications



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Thank you

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